

International Union of Crystallography

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Prices of *Acta Crystallographica* and *Journal of Applied Crystallography*

The Executive Committee of the International Union of Crystallography is pleased that it has not been necessary to increase the subscription rates and the prices of back numbers for *Acta Crystallographica* and *Journal of Applied Crystallography* as from 1 January 1987.

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A few copies of the cumulative index for Volumes 1–10 (1948–1957) are also available, free of charge.

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Vols. 1–19	Dkr 400
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Orders

Orders for *Acta Crystallographica* and *Journal of Applied Crystallography* may be addressed to Munksgaard International Publishers Ltd, 35 Nørre Søgade, DK-1370 Copenhagen K, Denmark. Orders from subscribers in North America may alternatively be placed through Polycrystal Book Service, PO Box 27, Western Springs, IL 60558, USA.

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**Allocation of Papers in Section C
of *Acta Crystallographica***

Section C of *Acta Crystallographica* will be divided from Volume 43 into the following formal categories for the convenience of its readers:

Inorganic compounds

Those not containing any organic (C_xH_y) or substituted organic ($C_xH_{y-1}F$, C_xF_y , etc.) moiety. Included are metal

elements, intermetallic compounds, non-metal elements, compounds of non-metals (excluding organic compounds as defined above), salts of inorganic acids, coordination complexes not containing organic moieties. Metal carbides, carbonates, cyanides, isocyanides, cyanates, isocyanates, thiocyanates are regarded as inorganic, as are coordination complexes with carbon present only as carbonyl, cyanide, cyanato etc. ligands.

Examples: Be, Gd₂Pt, C(graphite), SiC, NaHCO₃,
K[Co(CN)(CO)₂(NO)].

Organometallic compounds

Those containing one or more metal atom and one or more organic (as defined above) moiety (not simply a metal-carbon bond). Included are metal salts of organic acids, coordination complexes in which carbon is present in an organic or substituted-organic ligand, metal(or metal salt)-organic addition compounds. Al, Ge, As, Sb and Po are defined as metals.

Examples: Ba(CHO₂)₂, K₄[Zr(C₂O₄)₄].5H₂O
(four oxalate ligands),
K[PtCl₃(C₂H₄)], CaBr₂.4CH₄O.

Organic compounds

Those containing carbon and hydrogen (or hydrogen partially or fully substituted by a non-metal) with or without additional non-metals. B, Si, Se, Te and At are defined as non-metals.

Examples: C₇H₄N₄, CCl₄ (carbon tetrachloride or tetra-
chloromethane),
C₆N₄ (ethylenetetra-carbonitrile), C₄H₈Se₄.

Biological macromolecules

Any substance containing a polypeptide, polynucleotide or polysaccharide chain longer than 25 residues.

Short-format papers will be divided into the same categories as for full papers.